

# Electronic, Magnetic, and Mechanical Properties of Fe<sub>3</sub>C and Fe<sub>7</sub>C<sub>3</sub> Carbides: A First-Principles Perspective

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## Introduction :

Iron carbides such as cementite (Fe<sub>3</sub>C) and Fe<sub>7</sub>C<sub>3</sub> play a critical role in various industrial applications, particularly in steel manufacturing and hard coatings due to their unique combinations of mechanical strength, magnetism, and stability [1, 2]. Understanding their fundamental properties is essential to optimizing performance and tailoring materials for specific applications. Density Functional Theory (DFT) has proven to be an effective tool for studying the atomic-level characteristics of such compounds [3]. In this work, first-principles calculations were performed within the framework of Density Functional Theory (DFT) using the Win2k Package [4]. The exchange-correlation interactions were treated using the Generalized Gradient Approximation (GGA) with the Perdew-Burke-Ernzerhof (PBE) functional. The projector augmented wave (PAW) method was employed to describe core-electron interactions. A plane-wave cutoff energy of 520 eV was used, and the Brillouin zone was sampled using Monkhorst-Pack grids of 7×7×7 and 5×5×7 for Fe<sub>3</sub>C and Fe<sub>7</sub>C<sub>3</sub>, respectively. Structural relaxations were performed until the forces on each atom were less than 0.01 eV/Å, and the energy convergence criterion was set to 10<sup>-6</sup> eV.

## Results and Discussion :

### *Structural Properties:*

The optimized lattice parameters for Fe<sub>3</sub>C and Fe<sub>7</sub>C<sub>3</sub> are consistent with experimental and previous theoretical values, confirming the reliability of the computational setup. The Fe<sub>3</sub>C crystallizes in an orthorhombic structure (Pnma), while Fe<sub>7</sub>C<sub>3</sub> adopts a hexagonal structure (P6<sub>3</sub>mc). The calculated equilibrium volumes and bond lengths suggest strong Fe–C interactions contributing to the structural stability.

### *Electronic Properties:*

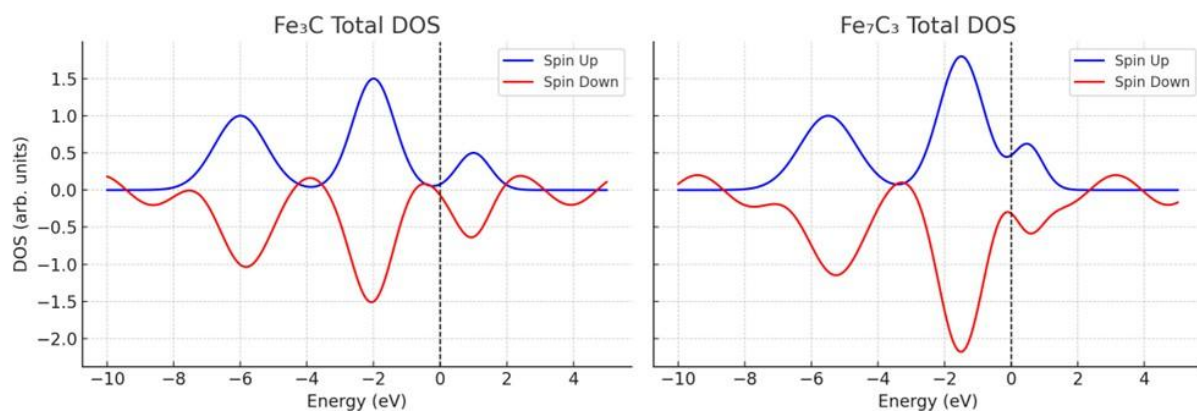
The electronic density of states (DOS) analysis shows metallic behavior for both carbides, with a significant contribution of Fe 3d states at the Fermi level. The presence of hybridization between Fe d and C p orbitals indicates covalent bonding characteristics in addition to metallic features. The DOS for  $\text{Fe}_7\text{C}_3$  exhibits a higher peak near the Fermi level, suggesting enhanced metallic conductivity.

#### *Magnetic Properties:*

compounds exhibit ferromagnetic ordering. The total magnetic moment per Fe atom is higher in  $\text{Fe}_7\text{C}_3$  than in  $\text{Fe}_3\text{C}$ , attributed to the lower symmetry and varying coordination environments in  $\text{Fe}_7\text{C}_3$ . Spin-polarized DOS reveals a clear exchange splitting in the Fe d bands, confirming strong ferromagnetic interactions.

#### *Mechanical Properties:*

Elastic constants were computed to evaluate mechanical stability and stiffness. Both  $\text{Fe}_3\text{C}$  and  $\text{Fe}_7\text{C}_3$  satisfy the Born mechanical stability criteria. Bulk modulus, shear modulus, and Young's modulus were derived using the Voigt-Reuss-Hill approximation.  $\text{Fe}_7\text{C}_3$  displays slightly higher stiffness, consistent with its denser atomic packing.



**Figure 1:** Total density of states (DOS) for  $\text{Fe}_3\text{C}$  and  $\text{Fe}_7\text{C}_3$

## **References :**

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